Response surface optimization for the continuous glucose isomerization process

WA Bazaraa¹ and EE Hassan²

¹Department of Food Science and Technology, College of Agriculture, Cairo University, Giza; ²Amriya Pharmaceutical Industries, Alexandria, Egypt

Production of fructose via a continuous glucose isomerization process was optimized using response surface methodology. Glucose isomerization was performed using immobilized glucose isomerase in a flow-through tubular reactor. Process factors eg pH (7.0–7.8), temperature (50–60°C), flow rate (5–17 ml min⁻¹), and glucose content (30–50% w/w) of the feedstock solution were simultaneously tested according to a central composite experimental design. Measured responses such as % isomerization, and fructose yield (g h⁻¹) had an excellent correlation with tested factors. The highest desirability, *D*, (geometric mean of % isomerization and fructose yield) was obtained when the feedstock (56–60°C) had 34–36% glucose, a pH of 7.4–7.8 and was pumped at 15 ml min⁻¹.

Keywords: immobilized glucose isomerase; response surface methodology; optimization

Introduction

Fructose is the sweetest natural sugar. Its sweetening index is about 2.34 and 1.73 times that of glucose and sucrose, respectively [17]. Therefore, enzymatic isomerization of glucose to fructose has received in recent years considerable attention as a commercial process. Immobilized glucose isomerase is currently used for the industrial production of high fructose syrup (HFS) which recently replaced sucrose in many food products [13]. Various techniques of immobilizing either partially purified glucose isomerase (GI) or cells were reported [3,4,7,9,11– 13,19,20,25]. Currently, several brands of immobilized GI products are commercially available for the production of HFS [26].

Steadily increasing labor and transportation costs, competition from different producers, as well as competition from sucrose itself may drive the HFS industry into a difficult situation. Therefore, reduction in production costs is required and can be achieved by optimizing process conditions. Glucose isomerization process optimization was previously reported [21,25]. Current literature did not reveal utilization of the response surface methodology (RSM) technique in optimizing the process of glucose isomerization.

Statistical optimization has become a common practice in biotechnology. Among the numerous optimization techniques, RSM is widely applied in developing many food, fermentation and drug processes [1,8,10,15,16,22,24]. Process optimization using RSM is usually achieved by simultaneous testing of numerous factors in a limited number of experiments. Therefore, RSM consumes less time and effort compared to the traditional, one-factor-at-a-time approach. Furthermore, RSM provides quantitative measurements of possible interactions between factors, difficult information to obtain using traditional optimization techniques. Detection and quantification of the interactions between various factors are of critical importance especially for optimizing the multivariate processes of food biotech nology. The objective of this work was to study simultaneously the effect of selected process factors eg glucose concentration, temperature, pH and flow rate of the feedstock on the continuous isomerization of glucose.

Materials and methods

Chemicals

Immobilized glucose isomerase (Sweetzyme T), having an activity of 350 IGIU g⁻¹, was provided by Novo Industri A/S (Bagsvaerd, Denmark). Anhydrous glucose was obtained from Fisher Scientific (Springfield, NJ, USA). Carbazol was purchased from Sigma Chemical Company (St Louis, MO, USA). Cysteine hydrochloride was obtained from Eastman Organic Chemicals (Rochester, NY, USA). All other reagents were of analytical grade.

Continuous isomerization of glucose

Continuous isomerization of glucose was performed using a flow-through tubular reactor. Eighty grams of immobilized glucose isomerase were packed into a 3.3 cm \times 27 cm reactor column (280-ml total volume and 130-ml void volume). Feedstock (FS) containing 0.24% MgSO₄ and desired glucose concentrations were pumped upward using a peristaltic pump at predetermined flow rates. The pH of the FS was adjusted with sodium carbonate whereas the temperature was controlled using a thermostated water-bath. The concentration of the fructose produced was determined using the modified cysteine-carbazol method [6,14].

Experimental design and data analysis

According to the central composite design [2], 27 experiments were performed to study the effects of the independent variables eg glucose concentration (x_1) , temperature (x_2) , pH (x_3) , and flow rate (x_4) of the FS on process

Correspondence: WA Bazaraa, Department of Food Science and Technology, College of Agriculture, Cairo University, Giza, Egypt Received 1 February 1996; accepted 12 July 1996

responses eg % isomerization (y_1) and fructose yield (y_2) so that:

 $y_1 = \frac{\text{amount of fructose produced}}{\text{initial amount of glucose}} \times 100$

$$y_2$$
 = amount of fructose produced h⁻¹

The range of independent variables is recorded in Table 1. An overall desirability function, D, was also calculated as the geometric mean of the two responses, ie

$$D = \sqrt{y_1 \cdot y_2}$$

The Statistical Analysis system (SAS) [18] was used to fit the second order polynomial equation to the experimental data shown in Tables 2 and 3. Measured responses were correlated to the studied factors using response surface regression procedures [18]. Further model simplification steps were performed using multiple regression and backward elimination procedures [18]. After determining the optimum process conditions using RSM, the optimum response (D) was experimentally verified under the optimum conditions. The response D at the optimum process conditions was compared to values predicted by the model.

Results and discussion

Table 3 summarizes the data obtained from the 27 experiments performed according to the central composite design. Glucose isomerization (%) ranged from 27.6 to 45.5% whereas fructose yield ranged from 55.5 to 154.8 g h⁻¹. Since values of each response alone may not be fully descriptive for the process performance, a desirability function, *D*, was generated as the geometric mean of both

 Table 1
 Codes for independent variables and their corresponding values investigated in the optimization process of glucose isomerization

Independent variable	Coded symbol	Levels		
		Coded	Uncoded	
Glucose (% w/w)	X ₁	+2	50.0	
		+1	45.0	
		0	40.0	
		-1	35.0	
		-2	30.0	
Temperature (°C)	X ₂	+2	60.0	
		+1	57.5	
		0	55.0	
		-1	52.5	
		-2	50.0	
рН	X ₃	+2	7.8	
	-	+1	7.6	
		0	7.4	
		$^{-1}$	7.2	
		-2	7.0	
Flow rate (ml min ⁻¹)	\mathbf{X}_4	+2	17.0	
		+1	14.0	
		0	11.0	
		-1	8.0	
		-2	5.0	

 Table 2
 Experimental design and the combination of independent variables used for the various treatments

Treatment No.	Glucose conc (%)	Temp (°C)	pH	Flow rate (ml min ⁻¹)
1	+1	+1	+1	+1
	+1	+1	+1	-1
2 3 4 5	+1	+1	-1	+1
4	+1	+1	-1	-1
5	+1	-1	+1	+1
6	+1	-1	+1	-1
7	+1	-1	-1	+1
8	+1	-1	-1	-1
9	-1	+1	+1	+1
10	-1	+1	+1	-1
11	-1	+1	-1	+1
12	-1	+1	-1	-1
13	-1	-1	+1	+1
14	-1	-1	+1	-1
15	-1	-1	-1	+1
16	-1	-1	-1	-1
17	+2	0	0	0
18	-2	0	0	0
19	0	+2	0	0
20	0	-2	0	0
21	0	0	+2	0
22	0	0	-2	0
23	0	0	0	+2
24	0	0	0	-2
25	0	0	0	0
26	0	0	0	0
27	0	0	0	0

 Table 3
 Response variable data obtained from optimization experiments

Treatment No.	Isomerization (%)	Fructose yield (g h ⁻¹)	Total desirability function
1	33.8	154.8	72.3
2	43.9	115.0	71.1
2 3	31.8	145.6	68.0
4	42.3	110.8	68.5
5	27.6	126.4	59.1
6	38.1	99.7	61.6
7	28.0	128.3	59.9
8	38.5	100.7	62.3
9	41.7	142.7	77.2
10	45.0	88.0	62.9
11	40.6	138.9	75.1
12	45.2	88.3	63.2
13	40.6	138.9	75.1
14	43.2	84.3	60.3
15	37.0	126.4	68.4
16	43.4	84.8	60.7
17	32.0	130.2	64.5
18	42.7	106.3	67.4
19	45.1	121.0	73.8
20	35.1	109.2	61.9
21	39.3	121.9	69.2
22	40.0	124.2	70.5
23	29.6	141.9	64.8
24	45.5	55.5	50.3
25	37.6	116.7	66.2
26	38.5	117.7	67.9
27	38.3	118.9	67.5

101

102

Parameter Parameter estimates Isomerization (%) Fructose yield Desirability function Fa $\mathbf{S}^{\mathbf{b}}$ F S F S 1013.7 1065.51 3057.14 Intercept 2778.11 1772.74 1377.00 -1.32-2.56-7.94-2.54 \mathbf{X}_1 -8.64-4.18-12.53-12.49 -11.83-10.25 -14.68-4.36 X_2 X_3 -162.67-177.41 -693.83 -631.48 -353.70-310.74 -1.78X₄ 0.19 2.48 _ $\mathbf{x_1}^2$ -0.010.02 _ 0.00 _ x_{2}^{2} 0.08 0.09 -0.04 0.06 X_3^2 9.96 11.81 42.81 41.32 21.90 20.12 x_4^2 -0.01 -0.49 -0.49 -0.24 -0.250.06 0.06 0.24 0.12 0.24 0.12 X_1X_2 -0.10 -0.32 -0.18 X_1X_3 -0.10 -0.10-0.30 -0.30-0.22 -0.22 X_1X_4 X_2X_3 0.25_ 0.96 0.48 0.09 0.01 0.25 0.25 0.08 X_2X_4 0.40 0.58 2.23 $x_3 x_4$ 2.22 1.14 1.39

Table 4 Parameter estimates for full second-order polynomials and simplified models correlating responses to independent variables

^aFull model.

^bSimplified model.

Contour plot No.	Glucose conc (%)	Temp (°C)	рН	Flow rate (ml min ⁻¹)
1	34-46	56-60	_	
2	32-36	-		15-17
3	_	56-60	_	13-15
4	-		7.4–7.8	15-17
Common range (optimum conditions)	34–36	56-60	7.4–7.8	15

responses. Desirability functions serve as a single parameter used when simultaneous evaluation of different responses is required [5]. Table 4 indicates parameter estimates for the full second-order polynomial equations and corresponding simplified models. In all cases, excellent correlations ($r^2 > 0.94$) were observed among responses and studied factors. However, some of the parameters in the full models were not highly significant. Therefore, response surface graphs of the reduced model of D were applied to determine the optimum process conditions (Table 5). Desirability function increased by increasing temperature with the best obtained results between 56 and 60°C. This was due to increased enzyme activity at higher temperatures. This observation agrees with previously reported [3,21,23] temperature values (50-60°C). Similarly, at a higher pH value (7.4-7.8) optimum D was observed. Although enzyme stability was not studied in this investigation, pH and temperature ranges were chosen to minimize enzyme deactivation [23]. Straatsma et al [21] reported a maximum activity of glucose isomerase at a pH of 7.65 [21]. On the other hand, glucose concentration and flow rate of the FS play a rather complicated role in the isomerization process. At low concentrations of glucose, % isomerization will be high due to availability of the enzyme. However, low glucose flow rate input will result in low fructose yield. More-

over, increased flow rate will decrease % isomerization due to low enzyme-substrate contact time. Optimum FS flow rate was 15 ml min⁻¹ whereas the optimum glucose concentration range was 32-46%, preferably 35%. This preferred value lies at the lower limit of substrate concentration recommended by the GI manufacturer (35-45%) [23]. Optimum conditions were presented in treatment No. 9 (Table 2). This treatment resulted in the highest D value as indicated in Table 3. Adequacy of the simplified model equation for predicting optimum response value was experimentally evaluated using a FS containing 35% glucose, pH of 7.6, temperature of 57.5°C and flow rate of 15 ml min⁻¹. This set of conditions was determined to be the optimum by the RSM optimization procedure. The experimental Dvalue (77.2) was very close to the predicted value (85.6) indicating a good predictor model.

In conclusion, statistical central composite optimization procedure and response surface methodologies were used to obtain equations that identified significantly contributing process factors. Response surface graphic analysis allowed optimal process factors to be located. The experimental data were highly comparable to the model-predicted results (actual D = 77.2, predicted D = 85.6). The obtained optimum conditions were in close agreement with those reported in the literature.

References

- 1 Badr HR and MK Hamdy. 1992. Optimization of acetone butanol production using response surface methodology. Biomass and Bioenergy 3: 45–55.
- 2 Bayne CK and IB Rubin. 1986. Practical Experimental Designs and Optimization Methods for Chemists. VCH Publishers, Deerfield, Florida.
- 3 Bazaraa WA and MK Hamdy. 1989. Fructose production by immobilized Arthrobacter cells. J Ind Microbiol 4: 267–274.
- 4 Chen FS, HS Weng and CL Lai. 1983. The performance of immobilized glucose isomerase supported by shrimp chitin in various types of reactors. Biotechnol Bioeng 25: 725–733.
- 5 Derringer G and R Suich. 1980. Simultaneous optimization of several response variables. J Qual Tech 12: 1214–1219.
- 6 Dische Z and E Borenfreund. 1951. A new spectrophotometric method for the detection and determination of keto sugars and trioses. J Biol Chem 192: 583–587.
- 7 Gaikwad SM and VV Deshpande. 1992. Immobilization of glucose isomerase on Indion 48-R. Enzyme Microbiol Technol 14: 855-858.
- 8 Hassan EE, RC Parish and JM Gallo. 1992. Optimized formulation of magnetic chitosan microspheres containing anticancer agent, oxantrazole. Pharm Res 9: 390–397.
- 9 Huitron C and J Limon-Lason. 1978. Immobilization of glucose isomerase to ion-exchange materials. Biotechnol Bioeng 20: 1377–1391.
- 10 Kemp T, M Karim, J Linden and R Tengerdy. 1989. Response surface optimization of *Lactobacillus plantarum* batch growth. Biotechnol Lett 11: 817–820.
- 11 Kumakura M and I Kaetsu. 1984. Immobilization of cells by radiation copolymerization of hydrophilic and hydrophobic monomers. Act Chim Hung 116: 345–351.
- 12 Lee YY, AR Fratzke, K Wung and GT Tsao. 1976. Glucose isomerase immobilized on porous glass. Biotechnol Bioeng 18: 389–413.
- 13 Linko YY, L Pohjola and P Linko. 1977. Entrapped glucose isomerase for high fructose syrup production. Process Biochem 12: 14–16.
- 14 Marshall RO and ER Kooi. 1957. Enzymatic conversion of D-glucose to D-fructose. Science 125: 648–649.
- 15 Mudahar GS, RT Toledo, JD Floros and JJ Jen. 1989. Optimization

of carrot dehydration process using response surface methodology. J Food Sci 54: 714–719.

- 16 Prapulla SG, Z Jaccob, N Chand, D Rajalakshmi and NG Karanth. 1992. Maximization of lipid production by *Rhodotoruli gracilis* CFR-1 using response surface methodology. Biotechnol Bioeng 40: 965–970.
- 17 Pritham GH. 1968. Anderson's Essentials of Biochemistry. p 74, Mosby, New York, NY.
- 18 SAS. 1988. SAS/STAT User's Guide, release 6.03 edn, pp 877–896. SAS Institute Inc, Cary, NC.
- 19 Schafhauser DY and KB Storey. 1992. Immobilization of glucose isomerase onto granular chicken bone. Appl Biochem Biotechnol 32: 79–87.
- 20 Shukla GL and KA Prabhu. 1988. Whole cell immobilization of pglucose isomerase enzyme on glass support. J Basic Microbiol 28: 457–461.
- 21 Straatsma J, K Vellenga, H De-Wilt and G Joosten. 1983. Isomerization of glucose to fructose. 2: Optimization of reaction conditions in the production of high fructose syrup by isomerization of glucose catalyzed by a whole cell immobilized glucose isomerase catalyst. Ind Eng Chem Process – Des Dev 22: 356–361.
- 22 Strobel Jr R and W Nakatsukasa. 1993. Response surface methods for optimizing *Saccharopolyspora spinosa*, a novel macrolide producer. J Ind Microbiol 11: 121–127.
- 23 Sweetzyme T Product Sheet. 1991. Novo Industri A/S, Bagsvaerd, Denmark.
- 24 Udeh K and B Achremowicz. 1993. Optimization of cultivation medium composition of an L-lysine producing mutant: the use of response surface methodology. Acta Microbiol Pol 42: 171–180.
- 25 Vasic-Racki D, N Pavlovic, S Cismek, M Drazic and B Husadzic. 1991. Development of reactor model for glucose isomerization catalyzed by whole-cell immobilized glucose isomerase. Bioprocess Eng 7: 183–187.
- 26 Verhoff FH, G Boguslawski, OJ Iantero, ST Schlager and YC Joa. 1985. Glucose isomerase. In: Comprehensive Biotechnology (Moo-Young M, ed), Vol 3, pp 837–859, Pergamon Press, New York.